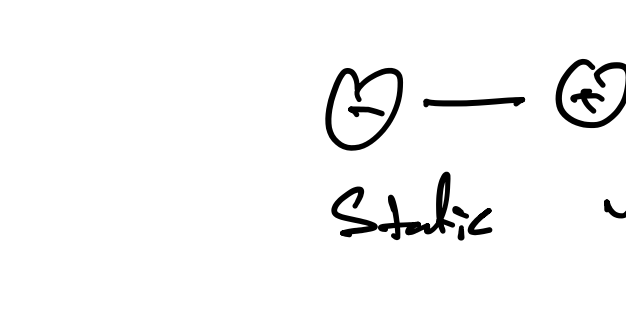
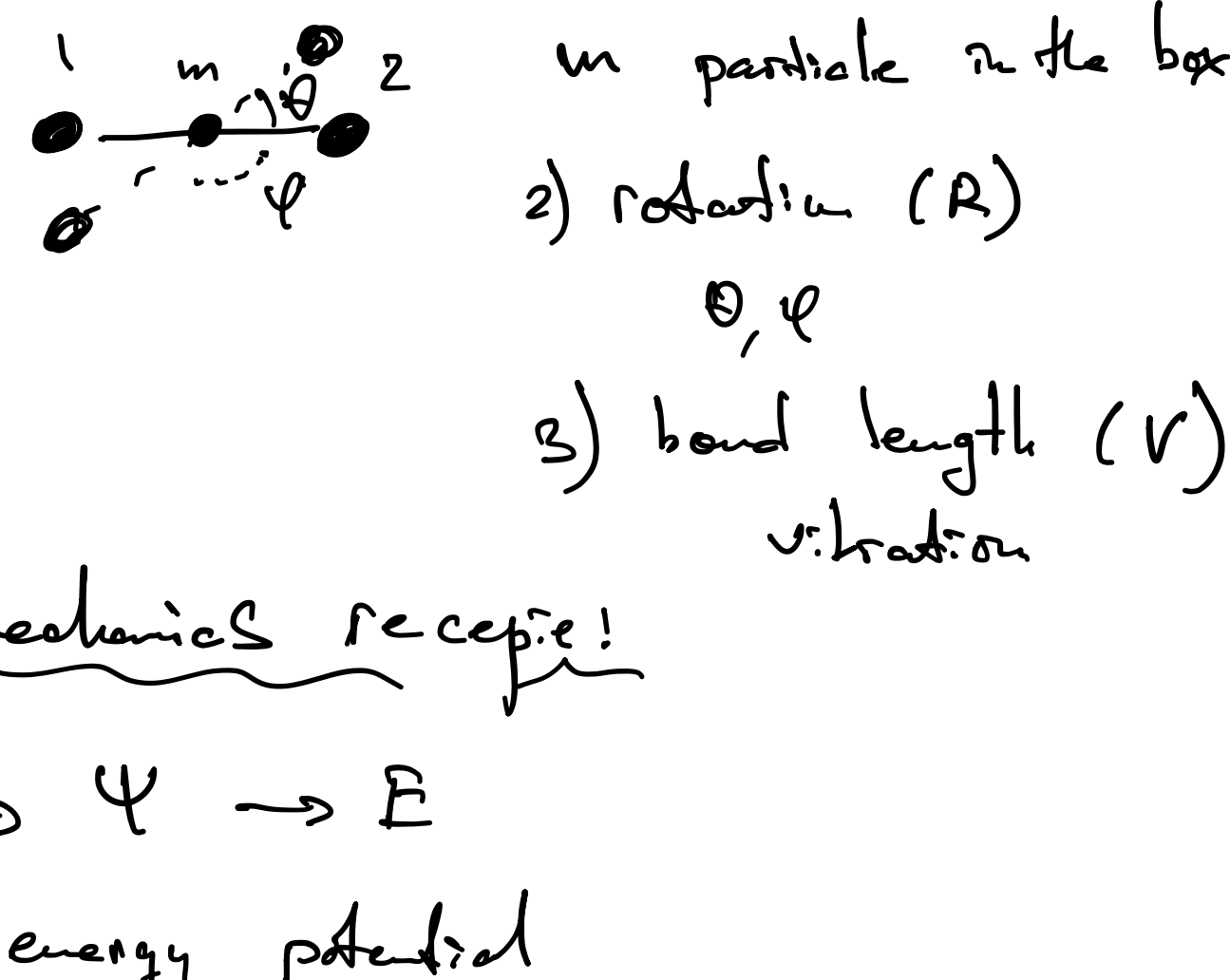


Idea of the dipole:



Electron or molecule

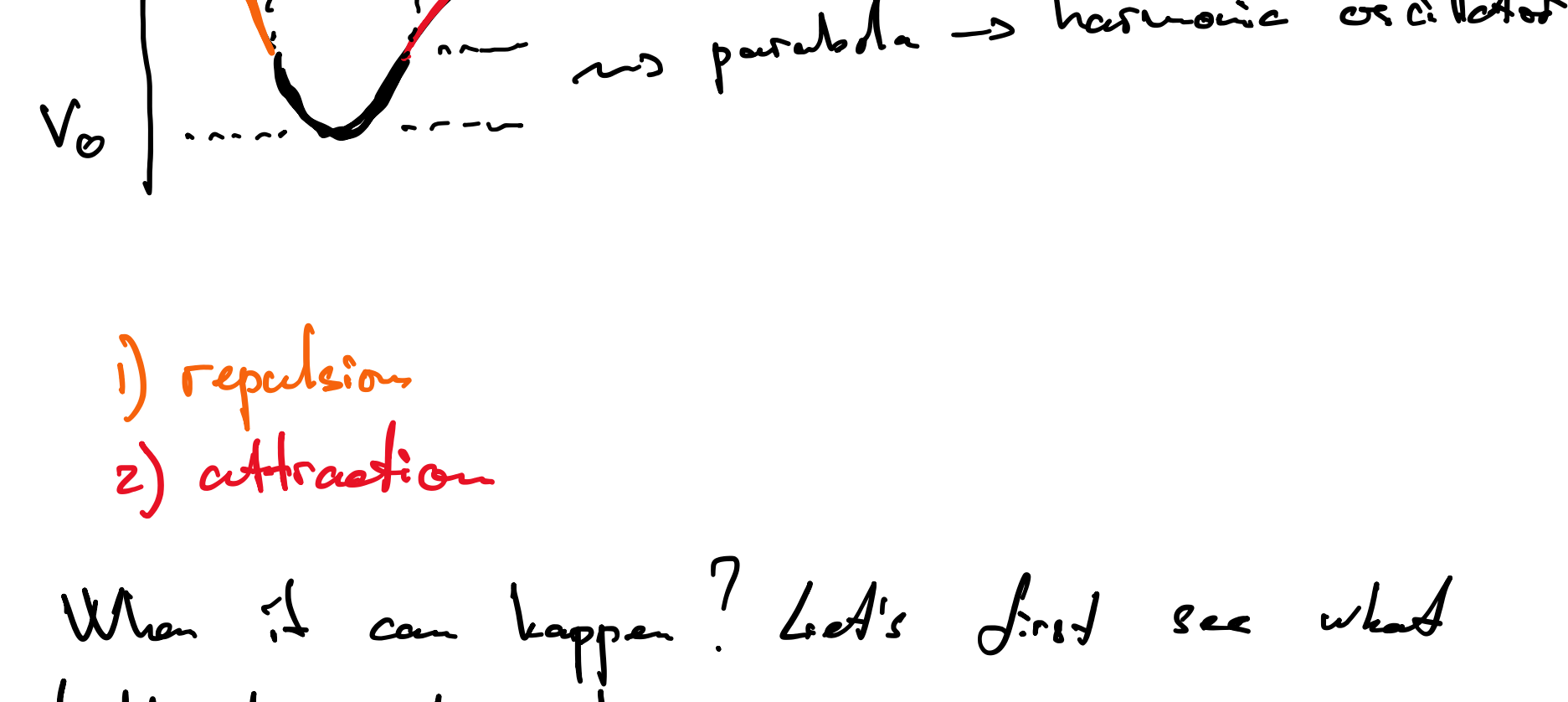


Quantum mechanics recipe!

$$V(r) \rightarrow \Psi \rightarrow E$$

- 1) establish energy potential
- 2) Solve Schrödinger equation
- 3) Find E.

$V(r)$ can be zero for T and R, but not for V



- 1) repulsion
- 2) attraction

When it can happen? Let's first see what fold the proton have:

$$e = 1.6 \cdot 10^{-19} \text{ C}$$

$$r_0 = 5.3 \cdot 10^{-11} \text{ m (Bohr radius)}$$

$$\epsilon_0 = 9 \cdot 10^{-12} \frac{\text{C}^2}{\text{V} \cdot \text{m}}$$

$$E = \frac{1}{4\pi\epsilon_0} \cdot \frac{e^2}{r_0^2} \sim 10^{12} \frac{\text{V}}{\text{m}}$$

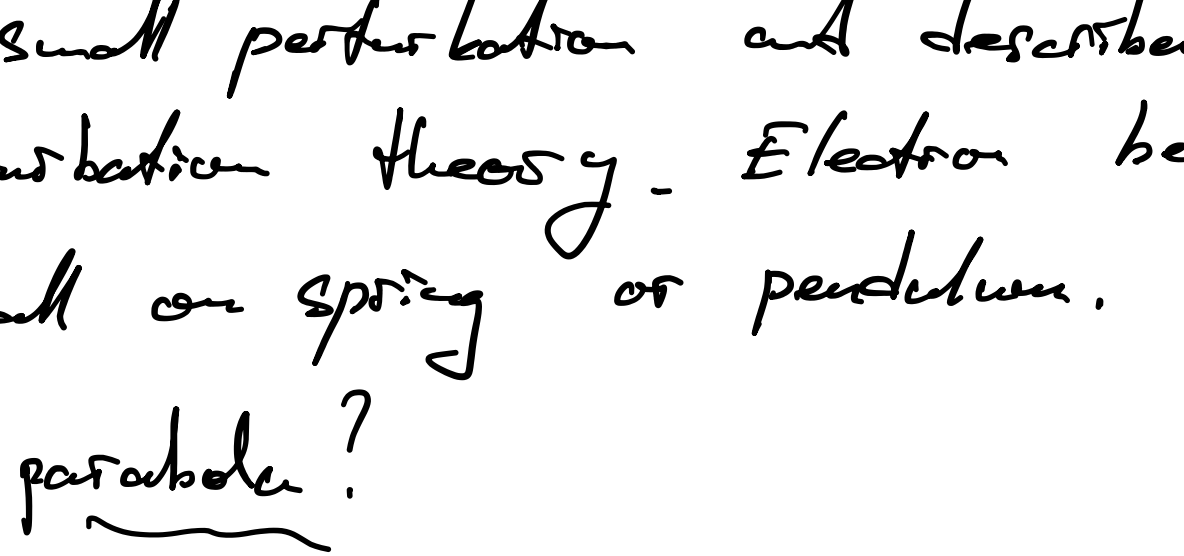
- ① Sun or HeNe
- ② fs pulse

$$I = 0.15 \frac{\text{W}}{\text{cm}^2}$$

$$I = 10^9 \frac{\text{W}}{\text{cm}^2} \text{ why?}$$

$$E = \sqrt{\frac{2I}{\epsilon_0 c}} = 3 \frac{\text{V}}{\text{m}}$$

$$E = \sqrt{\frac{2I}{\epsilon_0 c}} = 10^9 \frac{\text{V}}{\text{m}}$$



That is small perturbation and described by simple perturbation theory. Electron behaves like a ball on spring or pendulum.

Why it is parabolic?

We take simple model of ball on a string:

$$F = -kr \text{ (Hooke's force)}$$

in $F' \Rightarrow$ linear optics.

$$V(r) = \frac{1}{2} \cdot k \cdot r^2 \text{ - harmonic oscillator}$$

Since potential is established, we can solve Schrödinger equation!

$$\hat{H}\Psi = E\Psi$$

$$\underbrace{-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi}_{\text{kinetic}} + \underbrace{V(x) \Psi}_{\text{potential}} = E \Psi \quad \left(\begin{array}{l} \text{Schrödinger equation is} \\ \text{energy conservation law} \end{array} \right)$$

$$\left\{ \begin{array}{l} -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi + \frac{1}{2} k x^2 \Psi = E \Psi \\ \text{Schrödinger equation for harmonic oscillator} \end{array} \right.$$

$\Psi(x) \rightarrow$ one dimensional wavefunction

Let's see what function $\Psi(x)$ can be:

- ① Polynomial?

$$\left. \begin{array}{l} \Psi(x) = ax^3 \\ \Psi'(x) = 3ax^2 \\ \Psi''(x) = 6ax \end{array} \right\} \Rightarrow Ax + Bx^5 = Cx^3 \text{ not possible}$$

- ② Exponential (trigonometric)?

$$\left. \begin{array}{l} \Psi = ae^{bx} \\ \Psi' = abe^{bx} \end{array} \right\} \Rightarrow Ae^{bx} + Bx^2 e^{bx} = C e^{bx} \text{ better, but not yet there}$$

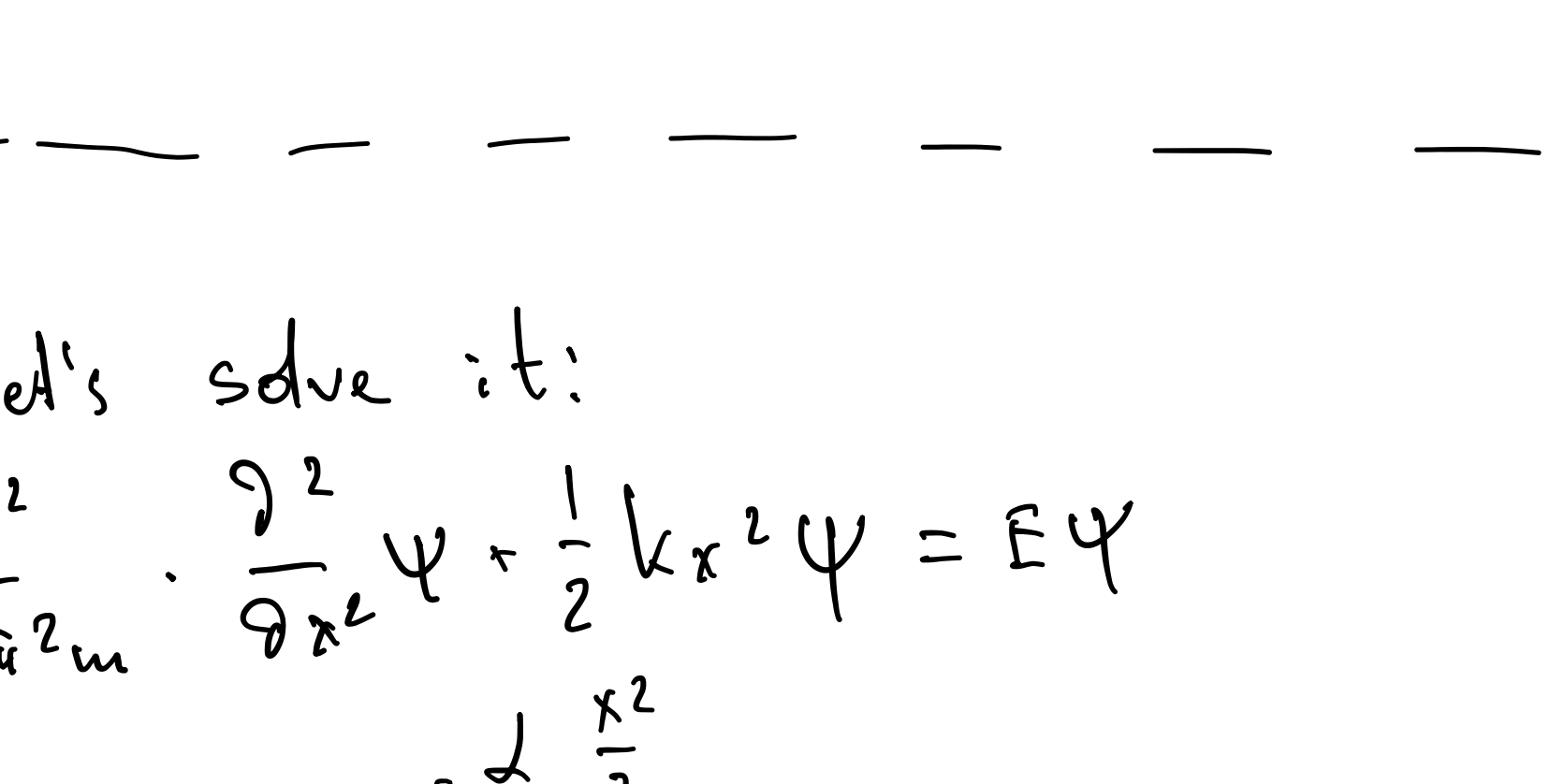
- ③ Combination

$$\left. \begin{array}{l} \Psi = ae^{bx^2} \\ \Psi' = 2abxe^{bx^2} \\ \Psi'' = 4abxe^{bx^2} + 4ab^2x^2 e^{bx^2} \end{array} \right\} \text{ can work!}$$

$$Ae^{bx^2} + Bx^2 e^{bx^2} + C x^2 e^{bx^2} = De^{bx^2}$$

Now, it is all about how we choose the coefficients.

How $\Psi(x)$ will look like?



- 1) Since $P = \Psi^2$ (probability to find particle) $b > 0 \Rightarrow \lim_{x \rightarrow \infty} P = \infty$
- 2) $b < 0 \Rightarrow$ Gauss $\Rightarrow \Psi(x) = N e^{-\alpha \frac{x^2}{2}}$ why?

Let's solve it:

$$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi + \frac{1}{2} k x^2 \Psi = E \Psi$$

$$\Psi(x) = N e^{-\alpha \frac{x^2}{2}}$$

$$\Psi(x) = N e^{-\alpha \frac{x^2}{2}}$$

$$\Psi' = -\alpha N x e^{-\alpha \frac{x^2}{2}}$$

$$\Psi'' = -\alpha N e^{-\alpha \frac{x^2}{2}} + \alpha^2 x^2 N e^{-\alpha \frac{x^2}{2}} = -\alpha^2 x^2 \Psi + \alpha \Psi$$

Hence:

$$-\frac{\hbar^2}{2m} (\alpha^2 x^2 - \alpha) \Psi + \frac{1}{2} k x^2 \Psi = E \Psi$$

Reorganize:

$$\left(-\frac{\hbar^2}{2m} \alpha^2 + \frac{1}{2} k \right) x^2 \Psi + \frac{\hbar^2}{2m} \alpha \Psi = E \Psi$$

This will only be possible if

$$\frac{1}{2} k - \frac{\hbar^2}{2m} \alpha^2 = 0$$

$$\alpha = \frac{\sqrt{2}}{\hbar} \sqrt{k m}$$

That also means:

$$E = \frac{\hbar^2}{2m} \alpha = \frac{\hbar^2}{2m} \cdot \frac{\sqrt{2}}{\hbar} \sqrt{k m} = \frac{1}{2} \cdot \frac{\hbar}{\sqrt{2}} \sqrt{k}$$

But k is a spring constant, m is reduced mass $\Rightarrow \omega_0 = \frac{1}{\sqrt{2}} \sqrt{\frac{k}{m}}$ - classical resonance

$$\left[E = \frac{1}{2} \hbar \omega_0 \right] - \text{ta-da-da!}$$

Is $\Psi(x) = N e^{-\alpha \frac{x^2}{2}}$ the only solution?

You can easily find:

$$\Psi_0 = N e^{-\alpha \frac{x^2}{2}}$$

$$\Psi_1 = N x e^{-\alpha \frac{x^2}{2}}$$

$$\Psi_2 = N (4\alpha x^2 - 2) e^{-\alpha \frac{x^2}{2}}$$

$\vdots$

$$\Psi_n(x) = N H_n(\alpha^{1/2} x) e^{-\alpha \frac{x^2}{2}}$$

$H_n(x)$ - Hermite polynomial

$$H_0(x) = 1 \text{ even}$$

$$H_1(x) = 2x \text{ odd}$$

$$H_2(x) = 4x^2 - 2 \text{ even}$$

$$H_3(x) = 8x^3 - 12x \text{ odd}$$

$\vdots$

$$H_n(x) = 2x H_{n-1}(x) - 2n H_{n-2}(x)$$

Normalization constants and orthogonality of wavefunctions.

$$\Psi_n(x) = N_n H_n(\alpha^{1/2} x) e^{-\alpha \frac{x^2}{2}}$$

$$\alpha = \frac{\sqrt{2}}{\hbar} \sqrt{k m}$$

Normalization is needed as total probability must be 1.

$$I = \int_{-\infty}^{+\infty} P(x) dx = \int_{-\infty}^{+\infty} \Psi^* \Psi dx$$

$$\Psi_0 = N_0 e^{-\alpha \frac{x^2}{2}}$$

$$I = \int_{-\infty}^{+\infty} N_0^2 e^{-\alpha \frac{x^2}{2}} e^{-\alpha \frac{x^2}{2}} dx = N_0^2 \int_{-\infty}^{+\infty} e^{-\alpha x^2} dx =$$

$$= N_0^2 \sqrt{\frac{\pi}{\alpha}} \Rightarrow N_0^2 = \sqrt{\frac{\alpha}{\pi}} \Rightarrow N_0 = \left(\frac{\alpha}{\pi} \right)^{1/4}$$

Ψ_n and Ψ_m have to be orthogonal,

$$\text{i.e. } \int \Psi_n \Psi_m dx = 0$$

That comes from parity of wavefunctions and Hermite polynomials.



$$\int \Psi_2 \Psi_0 = 0 \text{ due to Hermite polynomials}$$